

ACYLATED FLAVONOLS FROM *CROCOSMIA CROCOSMIIFLORA*

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Abstract—Two new acylated flavonols, montbretins A and B were isolated from the corms of *Crococsmia crocosmiflora*. Their structures were established as myricetin 3-*O*-[β -D-glucopyranosyl(1 \rightarrow 2)-6-*O*-caffeoyl- β -D-glucopyranosyl(1 \rightarrow 2)- α -L-rhamnopyranoside]-4'-*O*-[α -L-rhamnopyranosyl(1 \rightarrow 4)- β -D-xylopyranoside] and 3-*O*-[β -D-glucopyranosyl(1 \rightarrow 2)-6-*O*-*p*-coumaroyl- β -D-glucopyranosyl(1 \rightarrow 2)- α -L-rhamnopyranoside]-4'-*O*-[α -L-rhamnopyranosyl(1 \rightarrow 4)- β -D-xylopyranoside] by spectral and chemical methods.

INTRODUCTION

Crococsmia crocosmiflora N.E.Br. (Montbretia) is well known as a garden plant and is used as an antitumour agent in Japanese folk medicine.

We report here the isolation and characterisation of two new acylated flavonols, named montbretins A(1) and B(2) from the corms of this plant.

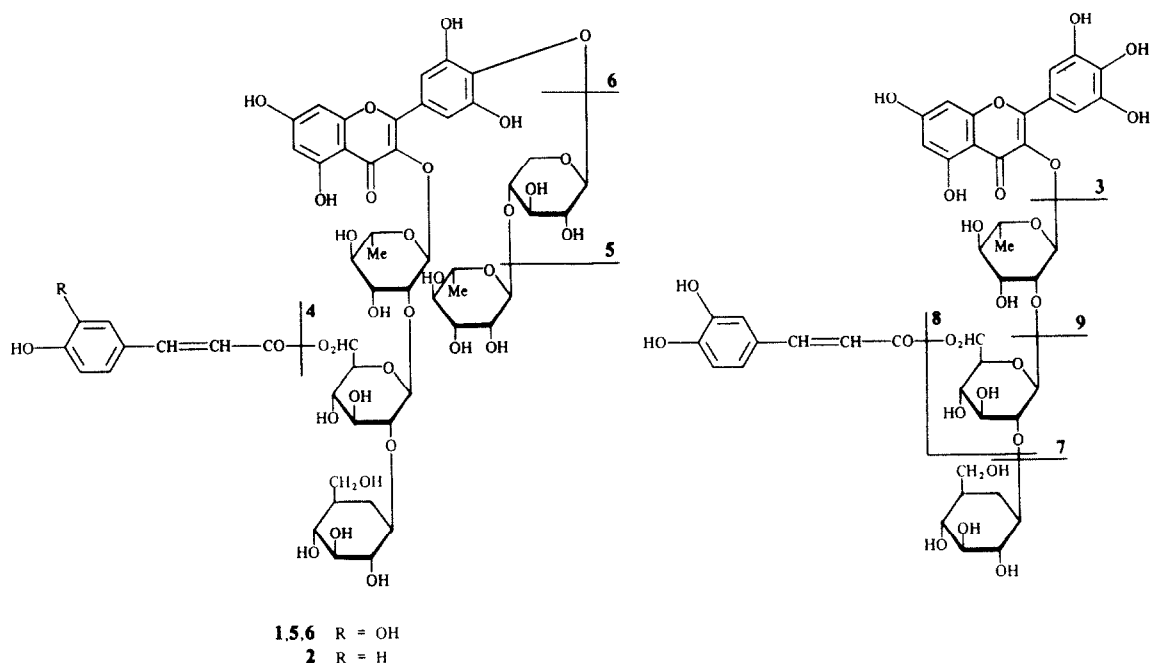
RESULT AND DISCUSSION

Montbretins A(1) and B(2) were isolated from the butanol-soluble fraction of a methanolic extract of *Crococsmia crocosmiflora* corms by a combination of silica gel and Sephadex LH-20 chromatography.

Montbretin A(1) was shown by elemental analysis and FABMS (m/z 1229[M+H] $^{+}$), to have the constitution $\text{C}_{53}\text{H}_{64}\text{O}_{33}$. It showed UV maxima at 265 and 326 nm and IR bands at 3400, 1695, 1660 and 1600 cm^{-1} . The ^1H NMR spectrum of 1 showed the presence of myricetin, a caffeoyl group and five sugar moieties. Myricetin was indicated by the appearance of three aromatic proton signals at δ 6.16 (1H, *d*, $J=2$ Hz), 6.21 (1H, *d*, $J=2$ Hz) and 6.91 (2H, *s*). The caffeoyl group was suggested by the appearance of ABC-type aromatic proton signals at δ 6.65 (1H, *d*, $J=8$ Hz), 6.74 (1H, *dd*, $J=8, 2$ Hz) and 6.85 (1H, *d*, $J=2$ Hz) and two *trans*-olefinic proton signals at δ 6.04 (1H, *d*, $J=16$ Hz) and 7.38 (1H, *d*, $J=16$ Hz). The appearance of five anomeric proton signals at δ 4.56 (1H, *d*, $J=8$ Hz), 4.61 (1H, *d*, $J=8$ Hz), 4.83 (1H, *br s*), 4.84 (1H, *d*, $J=7$ Hz) and 5.66 (1H, *br s*) showed the presence of five sugar moieties. Furthermore, the appearance of five anomeric carbon signals at δ 99.9, 102.6, 105.4, 106.5 and 107.1 in the ^{13}C NMR spectrum of 1 suggested the presence of five monosaccharide units. Acid hydrolysis of 1 with 4% sulphuric acid gave myricetin (3) [1], glucose, rhamnose and xylose (2:2:1) by HPLC and TLC. Alkaline hydrolysis of 1 with 4% potassium hydroxide yielded descaffeoylmontbretin A(4) and caffeic acid. The structure of 4 was confirmed by a comparison of its IR and ^1H NMR spectra with those of 1, which showed that the band at 1695 cm^{-1} in the IR

spectrum and signals for the caffeoyl group in the ^1H NMR spectrum of 4 had disappeared. In the ^1H NMR spectrum of 4, the C_6 -proton signals of glucose moiety were shifted to a higher field compared with those of 1 [2, 3]. Furthermore, in the ^{13}C NMR spectra, acylation shifts [4, 5] of glucose carbons for 1, compared with 4 were observed at C-4 (+1.0 ppm), C-5 (−2.4 ppm) and C-6 (+2.0 ppm) (Table 1). These results indicated that the caffeoyl group may be attached to the C_6 -position of a glucose in 1.

In order to elucidate the linkage positions and glycosidic linkage between the constituents in 1, enzymatic hydrolyses were performed. Thus, hydrolysis of 1 with naringinase [6] for 1 hr yielded the hydrolysate 5 and rhamnose. In addition, the assignable methyl proton and carbon signals of rhamnose moiety present in the ^1H and ^{13}C NMR spectra of 1 had disappeared in the correspondent spectra of 5. The linkage of the rhamnose in 1 to the C_4 position of the xylose moiety was decided on the basis of the ^{13}C NMR spectra of 1 and 5 which showed glycosylation shifts [7, 8] of the xylose carbons for 1, compared with 5 at C-3 (−1.6 ppm), C-4 (+4.2 ppm) and C-5 (−2.9 ppm) (Table 1). On enzymatic hydrolysis with naringinase for 4 days, 5 gave a hydrolysate (6) and xylose. The hydrolysate (6) was also prepared directly from 1. The ^{13}C NMR spectrum of 6 showed the disappearance of the xylose moiety when compared with that of 5. The glycosylation shifts [9, 10] of aglycone carbons for 5, compared with 6 were observed at C-1' (+6.6 ppm), C-3',5' (+4.9 ppm) and C-4' (−0.4 ppm) suggesting that the xylose moiety in 5 was linked to the 4'-hydroxyl group of the aglycone (myricetin) (Table 2). Comparison of the ^{13}C NMR spectra of 6 with myricetin (3) revealed glycosylation shifts of the aglycone carbons for 6 at C-2 (+10.3 ppm), C-3 (−0.3 ppm) and C-4 (+2.2 ppm) indicating that the three sugars in 6 were attached to the 3-position of myricetin. Mild enzymatic hydrolysis of 6 with hesperidinase [11, 12] for 26 hr. gave caffeic acid, hydrolysates 7 and 8 and myricetin(9) [10, 13]. The structure of 7 was confirmed by analysing the ^{13}C NMR spectra of 6 and 7. Thus, the ^{13}C NMR spectrum of 7 showed the disappearance of a glucose

Table 1. ^{13}C chemical shifts and assignments of sugar moieties of montbretins A(1), B(2) and related compounds (CD_3OD)

			1	2	4	5	6	7	8	9
3'-O-Glycoside	Rha	1	102.6	102.5	102.6	102.7	102.7	102.8	102.8	103.7
		2	84.2	84.3	83.7	84.3	84.5	83.7	83.0	72.0
		3	72.1 ^a	72.1 ^a	72.3 ^a	72.2	72.3	71.9	72.0	72.0
		4	73.9	73.9	74.0	74.0	74.1	73.8	73.8	73.5
		5	71.8	71.8	71.8	71.9	71.8	71.9	71.9	72.1
		6	17.7	17.8	17.8	17.8	17.7	17.7	17.7	17.7
	Glc	1	105.4	105.4	105.6	105.5	105.6	107.3	107.2	
		2	84.6	84.6	85.2	84.7	84.7	75.4	75.4	
		3	77.7 ^b	77.7 ^b	77.9 ^b	77.8 ^b	77.7 ^b	77.7	77.8	
		4	71.4	71.6	70.4	71.4	71.0	71.6	70.8	
		5	75.3 ^c	75.3 ^c	77.7	75.4	75.3	75.3	77.9	
		6	64.1	64.3	62.1	64.2	63.9	64.2	62.2	
	Glc	1	106.5	106.5	106.7	106.5	106.5			
		2	75.9	75.9	76.0	76.0	76.0			
		3	79.1	79.1	79.2	79.2	79.2			
		4	70.8	70.9	70.9	70.8	71.0			
		5	77.8 ^b	77.8 ^b	77.7 ^b	77.8 ^b	77.9 ^b			
		6	62.4	62.4	62.4	62.4	62.4			
4'-O-Glycoside	Xyl	1	107.1	107.2	107.2	107.3				
		2	74.8	74.9	74.8	74.7				
		3	75.6	75.6	75.6	77.2				
		4	75.2 ^c	75.2 ^c	75.3	71.0				
		5	64.5	64.6	64.5	67.4				
	Rha	1	99.9	99.9	99.9					
		2	72.1 ^a	72.1 ^a	72.2 ^a					
		3	72.4 ^a	72.4 ^a	72.4 ^a					
		4	73.9	73.9	74.0					
		5	70.1	70.1	70.2					
		6	18.0	18.0	18.0					

^{a-c} Assignments may be reversed in each columnRha: α -L-rhamnopyranosyl, Glc: β -D-glucopyranosyl, Xyl: β -D-xylopyranosyl

Table 2 ^{13}C chemical shifts and assignments of the aglycone of montbretins A(1) and B(2) and related compounds (CD_3OD)

		1	2	3	4	5	6	7	8	9
Aglycone	2	157.9	157.7	148.3	158.6	157.9	158.6	158.4	159.4	159.7
	3	136.1	136.1	137.2	136.2	136.1	136.9	136.9	136.7	136.5
	4	179.6	179.6	177.6	179.6	179.5	179.8	179.8	179.8	179.9
	5	163.0	163.1	162.7	163.2	163.2	163.1	163.2	163.2	163.3
	6	100.0	100.0	99.5	100.1	100.2	99.9	99.8	100.0	100.0
	7	166.0	165.9	165.8	166.2	166.4	165.9	165.7	166.0	166.0
	8	95.0	95.0	94.6	95.0	95.1	94.9	94.8	94.8	94.9
	9	158.4	158.3	158.5	158.7	158.5	158.7	158.7	158.6	158.7
	10	106.1	106.1	104.7	106.1	106.0	106.0	106.0	106.0	106.1
	1'	128.7	128.6	123.4	128.6	128.7	122.1	122.1	122.0	122.2
	2'	109.8	109.8	108.9	109.8	109.8	109.5	109.6	109.6	109.5
	3'	151.7	151.7	146.9	152.0	151.8	146.9	146.9	147.0	147.0
	4'	137.3	137.4	137.9	137.1	137.4	137.8	137.8	137.9	138.0
	5'	151.7	151.7	146.9	152.0	151.8	146.9	146.9	147.0	147.0
	6'	109.8	109.8	108.9	109.8	109.8	109.5	109.6	109.6	109.5
Caffeic acid and <i>p</i> -coumaric acid	1	127.6	127.0			127.6	127.7	127.8		
	2	115.1	131.0			115.1	115.0	115.1		
	3	149.4	116.7			149.5	149.4	149.4		
	4	146.6	161.0			146.7	146.7	146.7		
	5	116.6	116.7			116.4	116.4	116.4		
	6	122.9	131.0			123.0	123.0	123.0		
	1'	168.9	168.9			168.9	169.0	169.1		
	2'	114.7	114.8			114.8	114.9	114.9		
	3'	147.1	146.7			147.0	147.1	147.1		

moiety when compared with that of **6** and the glycosylation shifts of glucose carbons for **6** were observed at C-1 (-1.7 ppm) and C-2 ($+9.3$ ppm) compared with the spectrum of **7** suggesting that the terminal glucose in **6** is linked to the C-2 position of the inner glucose. The ^1H and ^{13}C NMR spectra of **8** showed the disappearance of the caffeoyl group as compared with those of **7**. The position of the caffeoyl group in **8** was decided from the acylation shift in the ^{13}C NMR spectrum. The acylation shifts of glucose carbons for **7** were observed at C-4 ($+0.8$ ppm), C-5 (-2.6 ppm) and C-6 ($+2.0$ ppm) as compared that of **8** suggesting that the caffeoyl group in **7** was attached at C-6 of the glucose. A comparison of the ^{13}C NMR spectra of **8** and **9** showed the glycosylation shifts of rhamnose carbons for **8** at C-1 (-0.9 ppm), C-2 ($+11.0$ ppm) and C-3 (-0.2 ppm) indicating the glucose in **8** was attached at C-2 of the rhamnose. On enzymatic hydrolysis with hesperidinase for 2 days, **6** yielded caffeic acid, hydrolysates **8**, myricetin and myricitrin.

With regard to the anomeric configuration of the glycosidic linkage, the β -orientations of the glucopyranoside and xylopyranoside were indicated by the corresponding anomeric proton signal patterns (d , $J = 7-8$ Hz) in the ^1H NMR spectrum of **1**, **4-8**. The anomeric configuration of the rhamnose in **1** was assigned as α by the application of Klyne's rule [14]. $[\text{M}]_D$ of **1** $-[\text{M}]_D$ of **6** = -355° . From these results, montbretin A(**1**) was characterised as myricetin 3-*O*-[β -D-glucopyranosyl (1 \rightarrow 2)-6-*O*-caffeoyl- β -D-glucopyranosyl (1 \rightarrow 2)- α -L-rhamnopyranoside]-4'-*O*-[α -L-rhamnopyranosyl (1 \rightarrow 4)- β -D-xylopyranoside].

The molecular formula of montbretin B(**2**) was confirmed as $\text{C}_{53}\text{H}_{64}\text{O}_{32}$ by elemental analysis and FABMS (m/z 1213 $[\text{M} + \text{H}]^+$). It gave UV maxima at

267, 304 and 311 nm and IR bands at 3380, 1685, 1650 and 1600 cm^{-1} . The ^1H NMR spectrum of **2** is similar to that of **1** except for the A_2B_2 -type aromatic proton signals at δ 6.67 (2H, d , $J = 8$ Hz) and 7.22 (2H, d , $J = 8$ Hz) suggesting the replacement of the caffeoyl group in **1** by a *p*-coumaroyl moiety in **2**. In the ^{13}C NMR spectrum of **2**, the resonances due to the aglycone and sugar carbons are quite coincident with that of **1** (Tables 1 and 2), suggesting that **2** has the same sugar components as **1** and that the *p*-coumaroyl group in **2** may be attached to the C-6 position of the inner glucose as the caffeoyl group is in **1**. Acid hydrolysis of **2** with 4% sulphuric acid gave myricetin (**3**), glucose, rhamnose and xylose (2:2:1). Alkaline hydrolysis of **2** with 4% potassium hydroxide yielded *p*-coumaric acid and the hydrolysate (**4**) which was identified as descaffeoylmontbretin A(**4**) obtained from **1**.

On the basis of chemical and spectroscopic evidence, montbretin B(**2**) was identified as myricetin 3-*O*-[β -D-glucopyranosyl (1 \rightarrow 2)-6-*O*-*p*-coumaroyl- β -D-glucopyranosyl (1 \rightarrow 2)- α -L-rhamnopyranoside]-4'-*O*-[α -L-rhamnopyranosyl (1 \rightarrow 4)- β -D-xylopyranoside].

Flavonoid pentaglycosides rarely occur in the plant kingdom and this is the first report of a 3,4'-diglycosylmyricetin with either a caffeoyl or *p*-coumaroyl moiety [1, 10, 15].

EXPERIMENTAL

Mps. uncorr. ^1H and ^{13}C NMR spectra were taken on JEOL FX-270 and Varian XL-300 spectrometers. EI and FABMS were taken with JEOL DX-300 instruments equipped with a direct inlet system.

Isolation of flavonoids The fresh corms (4.5 kg) of *C. crocosmiflora* NE Br. were extracted with MeOH. The concd methanolic extract was extracted successively with *n*-hexane, C_6H_6 , EtOAc and *n*-BuOH. The *n*-BuOH extract was repeatedly chromatographed on silica gel with the upper layer of *n*-BuOH-EtOAc- H_2O (5:1:4) to give the fraction containing **1** and **2** (Fr. 1). The repeated separation of Fr. 1 by Sephadex LH-20 chromatography gave **1** (634 mg) and **2** (155 mg). Montbretin A(1) Yellow powder $[\alpha]_D^{25} -42.1$ (H_2O , c 1.23). Anal. calcd for $C_{25}H_{34}O_{13}$: C, 51.78%, H, 5.25%. Found: C, 51.76%, H, 5.35%. UV λ_{max}^{MeOH} nm 265, 326. FABMS m/z 1229 ($M+1$)⁺. ¹H NMR (CD_3OD) δ 1.07 (3H, *d*, $J=6$ Hz), 1.28 (3H, *d*, $J=6$ Hz), 4.20 (1H, *dd*, $J=11$, 5 Hz), 4.48 (1H, *d*, $J=11$ Hz), 4.56 (1H, *d*, $J=8$ Hz), 4.61 (1H, *d*, $J=8$ Hz), 4.81 (1H, *br s*), 4.86 (1H, *d*, $J=7$ Hz), 5.66 (1H, *br s*), 6.04 (1H, *d*, $J=16$ Hz), 6.16 (1H, *d*, $J=2$ Hz), 6.21 (1H, *d*, $J=2$ Hz), 6.65 (1H, *d*, $J=8$ Hz), 6.74 (1H, *dd*, $J=8$, 2 Hz), 6.85 (1H, *d*, $J=2$ Hz), 6.91 (2H, *s*), 7.38 (1H, *d*, $J=16$ Hz).

Montbretin B(2) Yellow powder $[\alpha]_D^{25} -71.1$ (EtOH, c 1.37). Anal. calcd for $C_{25}H_{34}O_{13}$: C, 52.46%, H, 5.32%. Found: C, 52.12%, H, 5.38%. UV λ_{max}^{MeOH} nm 267, 304 (sh), 326. FABMS m/z 1213 ($M+1$)⁺. ¹H NMR (CD_3OD) δ 1.08 (3H, *d*, $J=6$ Hz), 1.29 (3H, *d*, $J=6$ Hz), 4.20 (1H, *dd*, $J=11$, 5 Hz), 4.48 (1H, *d*, $J=11$ Hz), 4.59 (1H, *d*, $J=8$ Hz), 4.61 (1H, *d*, $J=8$ Hz), 4.83 (1H, *br s*), 4.83 (1H, *d*, $J=7$ Hz), 5.69 (1H, *br s*), 6.05 (1H, *d*, $J=16$ Hz), 6.16 (2H, *s*), 6.67 (2H, *d*, $J=8$ Hz), 6.89 (2H, *s*), 7.22 (2H, *d*, $J=8$ Hz), 7.42 (1H, *d*, $J=16$ Hz).

Acid hydrolysis of 1. A soln of **1** (51.8 mg) in 4% H_2SO_4 (2 ml) was heated at 100° for 3.5 hr giving a ppt of myricetin (3.45 mg), ident. by comparison with an authentic marker. The filtrate was passed through Amberlite IRA-45 (OH form) and concd to give the sugar fraction. Sugars were identified by TLC [silica gel, solvent $CHCl_3$ -MeOH- H_2O (6:4:1)] and HPLC [column ERC-NH-1171 (6 \times 200 mm), solvent 80% CH_3CN , flow rate 1.5 ml, detector RI, R_f (min): rhamnose (4.7), xylose (5.4), glucose (7.4)].

Alkaline hydrolysis of 1. A soln of **1** (143 mg) in 4% KOH (14 ml) was stood for 2 hr at room temp. and then acidified with 0.2 N HCl. The acidic soln was adsorbed by Diaion HP-20 and eluted successively with H_2O , 60% MeOH and 80% MeOH. The 60% MeOH soln was evapd and the residue chromatographed on a silica gel with $CHCl_3$ -MeOH- H_2O (7:3:0.5) to give caffeic acid (9.7 mg) and descaffeoylmontbretin A(4, 86.0 mg). **4** Yellow powder $[\alpha]_D^{25} -73.3$ (H_2O , c 1.29). UV λ_{max}^{MeOH} nm 269, 311, 337 (sh). IR ν_{max}^{KBr} cm^{-1} 3400, 1655, 1600. ¹H NMR (CD_3OD) δ 0.98 (3H, *d*, $J=6$ Hz), 1.28 (3H, *d*, $J=6$ Hz), 4.49 (1H, *d*, $J=8$ Hz), 4.58 (1H, *d*, $J=8$ Hz), 4.83 (1H, *br s*), 4.84 (1H, *d*, $J=8$ Hz), 5.62 (1H, *br s*), 6.20 (1H, *d*, $J=2$ Hz), 6.32 (1H, *d*, $J=2$ Hz), 6.92 (2H, *s*).

Enzymatic hydrolysis of 1 with naringinase. A soln of **1** (21 mg) and naringinase (22 mg) in 10% EtOH (4 ml) was incubated at 37° for 1 hr. The reaction mixture was diluted with H_2O (25 ml) and extracted with *n*-BuOH. The concd BuOH soln was separated by CC on Sephadex LH-20 (MeOH), affording **5** (10 mg). **5** Yellow powder $[\alpha]_D^{25} -15.0$ (H_2O , c 0.71). UV λ_{max}^{MeOH} nm 256 (sh), 265, 305 (sh), 326. IR ν_{max}^{KBr} cm^{-1} 3350, 1685, 1650, 1600. ¹H NMR (CD_3OD) δ 1.07 (3H, *d*, $J=6$ Hz), 4.55 (1H, *d*, $J=8$ Hz), 4.58 (1H, *d*, $J=8$ Hz), 4.82 (1H, *d*, $J=7$ Hz), 5.65 (1H, *br s*), 6.05 (1H, *d*, $J=16$ Hz), 6.17 (1H, *d*, $J=2$ Hz), 6.21 (1H, *d*, $J=2$ Hz), 6.66 (1H, *d*, $J=8$ Hz), 6.75 (1H, *dd*, $J=8$, 2 Hz), 6.85 (1H, *d*, $J=2$ Hz), 6.91 (2H, *s*), 7.39 (1H, *d*, $J=16$ Hz). The residual soln was evapd to give rhamnose ident. by TLC and HPLC with an authentic sample.

Enzymatic hydrolysis of 1 with naringinase. A soln of **1** (26 mg) and naringinase (56 mg) in 10% EtOH (4 ml) was incubated at 37° for 4 days. The hydrolysate was treated just in the same

manner as in the case of **5**, affording **6** (12.8 mg). **6** Yellow powder $[\alpha]_D^{25} +41.4$ (H_2O , c 1.06). Anal. calcd for $C_{42}H_{46}O_{25}$: C, 53.04%, H, 4.88%. Found: C, 52.51%, H, 5.07%. UV λ_{max}^{MeOH} nm 249, 263 (sh), 304 (sh), 332. IR ν_{max}^{KBr} cm^{-1} 3340, 1695, 1660, 1600. FABMS m/z 951 ($M+1$)⁺. ¹H NMR (CD_3OD) δ 1.09 (3H, *d*, $J=6$ Hz), 4.34 (1H, *d*, $J=8$ Hz), 4.61 (1H, *d*, $J=8$ Hz), 5.48 (1H, *br s*), 6.10 (1H, *d*, $J=16$ Hz), 6.17 (1H, *d*, $J=2$ Hz), 6.24 (1H, *d*, $J=2$ Hz), 6.68 (1H, *d*, $J=8$ Hz), 6.79 (1H, *dd*, $J=8$, 2 Hz), 6.90 (1H, *d*, $J=2$ Hz), 6.99 (2H, *s*), 7.41 (1H, *d*, $J=16$ Hz).

Enzymatic hydrolysis of 5 with naringinase. A soln of **5** (60 mg) and naringinase (124 mg) in 10% EtOH (10 ml) was incubated at 37° for 4 days, affording **6** (33 mg).

Enzymatic hydrolysis of 6 with hesperidinase. A soln of **6** (153 mg) and hesperidinase (76 mg) in H_2O (30 ml) was incubated at 37° for 26 hr. The reaction mixture was filtered through a column of Diaion HP-20 and eluted with 80% MeOH. The concd MeOH soln was separated by CC on Sephadex LH-20 (MeOH) and silica gel ($CHCl_3$ -MeOH- H_2O , 7:3:0.5) to give caffeic acid (14 mg), **7** (10 mg), **8** (30 mg), and myricitrin (**9**, 8 mg). **7** Yellow powder $[\alpha]_D^{25} -98.8$ (EtOH, c 0.58). UV λ_{max}^{MeOH} nm 249, 262 (sh), 303 (sh), 332. IR ν_{max}^{KBr} cm^{-1} 3400, 1695, 1660, 1605. ¹H NMR (CD_3OD) δ 1.08 (3H, *d*, $J=6$ Hz), 4.36 (1H, *d*, $J=8$ Hz), 5.58 (1H, *br s*), 6.08 (1H, *d*, $J=16$ Hz), 6.15 (1H, $J=2$ Hz), 6.22 (1H, *d*, $J=2$ Hz), 6.67 (1H, *d*, $J=8$ Hz), 6.78 (1H, *dd*, $J=8$, 2 Hz), 6.89 (1H, *d*, $J=2$ Hz), 6.97 (2H, *s*), 7.34 (1H, *d*, $J=16$ Hz). **8** Yellow powder $[\alpha]_D^{25} -130.7$ (EtOH, c 1.17). UV λ_{max}^{MeOH} nm 256, 306 (sh), 353. IR ν_{max}^{KBr} cm^{-1} 3380, 1655, 1605. ¹H NMR (CD_3OD) δ 1.02 (3H, *d*, $J=6$ Hz), 4.34 (1H, *d*, $J=7$ Hz), 5.56 (1H, *br s*), 6.20 (1H, *d*, $J=2$ Hz), 6.37 (1H, *d*, $J=2$ Hz), 6.97 (2H, *s*). Compound **9** was identical with an authentic sample of myricitrin on the basis of mixed fusion and spectral data.

Enzymatic hydrolysis of 6 with hesperidinase. A soln of **6** (153 mg) and hesperidinase (158 mg) in H_2O (30 ml) was incubated at 37° for 2 days and was filtered through a Diaion HP-20 column. From the concd filtrate glucose and rhamnose were ident. by TLC and HPLC comparison with authentic samples. Further elution of the HP-20 column with MeOH and separation of the concd eluate by CC on Sephadex LH-20 (MeOH) and silica gel ($CHCl_3$ -MeOH- H_2O , 5:5:1) gave myricetin (3, 13 mg), caffeic acid (9 mg), **8** (25 mg), myricitrin (**9**, 7 mg).

Acid hydrolysis of 2. A soln of **2** (10.3 mg) in 2% H_2SO_4 (2 ml) heated at 100° for 2 hr gave **3** (1.7 mg). The filtrate was passed through Amberlite IRA-45 (OH form) and concentrated to give a sugar fraction. Sugars were identified by TLC and HPLC as the same method used for **1**.

Alkaline hydrolysis of 2. A soln of **2** (21.4 mg) in 4% KOH (3 ml) was stood for 2.5 hr at room temp. and then acidified with 0.2 N HCl. The acidic soln was adsorbed on Diaion HP-20 and successively eluted with H_2O , 20% MeOH and 60% MeOH. The concd 60% MeOH soln was separated by CC on silica gel ($CHCl_3$ -MeOH- H_2O = 7:3:1, lower phase and $CHCl_3$ -MeOH- H_2O = 6:4:1) to give **4** (8.5 mg) and *p*-coumaric acid (0.4 mg, ident. by IR, MS, ¹H NMR and mp). Compound **4** was identified as descaffeoylmontbretin A (**4**) by direct comparison with **4** obtained from **1**.

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